

# **EXHIBIT R**

Confidential - Subject to Protective Order

Page 294

IN THE UNITED STATES DISTRICT COURT  
FOR THE SOUTHERN DISTRICT OF WEST VIRGINIA  
CHARLESTON DIVISION

- - -

IN RE: ETHICON, INC. PELVIC :MDL NO. 2327  
REPAIR SYSTEM, PRODUCTS :  
LIABILITY LITIGATION :VOLUME II  
:

THIS DOCUMENT RELATES TO ALL CASES AND  
VARIOUS OTHER CROSS-NOTICED ACTIONS

CONFIDENTIAL - SUBJECT TO PROTECTIVE ORDER

- - -

January 8, 2014

- - -

Transcript of the continued deposition of  
THOMAS A. BARBOLT, Ph.D., called for Videotaped  
Examination in the above-captioned matter, said  
deposition taken pursuant to Superior Court Rules of  
Practice and Procedure by and before Michelle L.  
Gray, a Certified Court Reporter, Registered  
Professional Reporter, and Notary Public, at the  
offices of Riker Danzig Scherer Hyland & Perretti  
LLP, Headquarters Plaza, One Speedwell Avenue,  
Morristown, New Jersey, commencing at 9:07 a.m.

- - -

GOLKOW TECHNOLOGIES, INC.  
877.370.3377 ph| 917.951.5672 fax  
deps@golkow.com

## Confidential - Subject to Protective Order

<p style="text-align: right;">Page 407</p> <p>1 BY MR. THORNBURGH:</p> <p>2 Q. IR spectra showed possible evidence</p> <p>3 of slight oxidation, correct?</p> <p>4 A. Yes.</p> <p>5 Q. Okay. Now, there's also an</p> <p>6 observation regarding the other Ethilon and Novofil,</p> <p>7 which differed from uncracked areas. And the</p> <p>8 conclusion was, expected IR absorbances for</p> <p>9 oxidation would be masked by strong carbonyl</p> <p>10 absorbances normally observed in these sutures.</p> <p>11 So there's a discussion here that --</p> <p>12 of the -- what would be expected to be seen could be</p> <p>13 masked by strong carbonyl absorbances. Do you see</p> <p>14 that?</p> <p>15 MR. THOMAS: Object to the form of</p> <p>16 the question.</p> <p>17 THE WITNESS: Yes.</p> <p>18 BY MR. THORNBURGH:</p> <p>19 Q. And at the seven-year data, Ethicon's</p> <p>20 investigator found degradation in Prolene is still</p> <p>21 increasing in PVDF -- even though a few cracks were</p> <p>22 found, is still by far the most surface resistant</p> <p>23 in-house made suture in terms of cracking.</p> <p>24 That's the findings by Ethicon's</p> <p>25 investigator, right?</p>	<p style="text-align: right;">Page 409</p> <p>1 BY MR. THORNBURGH:</p> <p>2 Q. And that's Ethicon's position as</p> <p>3 you -- as the spokesperson for Ethicon, it's</p> <p>4 Ethicon's position that degradation, surface</p> <p>5 degradation, can occur, correct?</p> <p>6 MR. THOMAS: Object to the form of</p> <p>7 the question.</p> <p>8 THE WITNESS: Yes.</p> <p>9 BY MR. THORNBURGH:</p> <p>10 Q. And this was known well in advance of</p> <p>11 this statement that the material is not absorbed,</p> <p>12 nor is it subject to degradation, correct?</p> <p>13 A. Yes. This is from 1992.</p> <p>14 MR. THORNBURGH: Okay. Lunch break.</p> <p>15 THE VIDEOGRAPHER: We're now going</p> <p>16 off the video record. It's 11:48.</p> <p>17 (Lunch break.)</p> <p>18 THE VIDEOGRAPHER: We're back on the</p> <p>19 video record. It's now 12:43.</p> <p>20 BY MR. THORNBURGH:</p> <p>21 Q. Now, Doctor, I'd like to turn your</p> <p>22 attention back to the e-mail that we began to</p> <p>23 discuss earlier in your deposition, Exhibit</p> <p>24 Number T 4012.</p> <p>25 (Whereupon, a discussion was held off</p>
<p style="text-align: right;">Page 408</p> <p>1 A. Yes.</p> <p>2 Q. An employee for Ethicon who actually</p> <p>3 investigated degradation of Prolene sutures and came</p> <p>4 to the conclusion that degradation is occurring in</p> <p>5 Prolene, right?</p> <p>6 MR. THOMAS: Object to the form of</p> <p>7 the question.</p> <p>8 BY MR. THORNBURGH:</p> <p>9 Q. Do you see that?</p> <p>10 A. Yes, I see that. Surface</p> <p>11 degradation, and they're making a reference to</p> <p>12 surface degradation. Yep. I see it.</p> <p>13 Q. So you agree as the person for</p> <p>14 Ethicon who's looked at these studies that surface</p> <p>15 degradation can occur on the Prolene polypropylene,</p> <p>16 correct?</p> <p>17 A. That was a surface change observed in</p> <p>18 this report and so reported.</p> <p>19 Q. And so you agree that surface</p> <p>20 degradation can occur in the Prolene polypropylene</p> <p>21 that's contained in the TVT meshes, correct?</p> <p>22 MR. THOMAS: Object to the form of</p> <p>23 the question.</p> <p>24 THE WITNESS: That's the data in this</p> <p>25 report reflecting the SEM parameters evaluated.</p>	<p style="text-align: right;">Page 410</p> <p>1 the record.)</p> <p>2 THE WITNESS: Okay.</p> <p>3 BY MR. THORNBURGH:</p> <p>4 Q. Now, this e-mail --</p> <p>5 MR. THOMAS: Give me just a half a</p> <p>6 second to get back on the same page.</p> <p>7 Thank you. I am ready.</p> <p>8 BY MR. THORNBURGH:</p> <p>9 Q. This e-mail is again from</p> <p>10 Dr. Divilio, and you were copied on this e-mail,</p> <p>11 right?</p> <p>12 A. Yes.</p> <p>13 Q. In 2007, correct?</p> <p>14 A. Yes.</p> <p>15 Q. And the e-mail says: Bruce Ramshaw</p> <p>16 from the University of Missouri is challenging our</p> <p>17 perception of polypropylene as an inert material</p> <p>18 after implantation. In a recent article, his group</p> <p>19 looked at explanted polypropylene from a Bard</p> <p>20 Composix mesh under EM, electron microscopy, and</p> <p>21 found that the surface of the fibers had been</p> <p>22 altered with respect to the pristine material with</p> <p>23 evidence of blistering and increased surface</p> <p>24 roughness, possibly due to oxidation.</p> <p>25 Now, this is the same finding or</p>

30 (Pages 407 to 410)